A SYNTHETIC APPROACH TO CYCLIC PEPTIDE MODELS BY REGIOSELECTIVE REMOTE PHOTOCYCLIZATION OF SULFIDE-CONTAINING PHTHALIMIDES 1,2

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Syntheses of cyclic peptides have received much attention because a number of biologically active derivatives including antibiotics and peptide hormones³ have been found. For synthesizing cyclic peptides it is generally essential to effect cyclization of linear peptide substrates with a selected activating group such as azide at high dilutions^{3a}. Our recent finding that certain sulfide-containing phthalimides undergo unusually facile and regional elective remote photocyclization leading to macrocycles⁴ prompted us to test the possibility of employing this technique for the synthesis of cyclic peptide analogs. In a preliminary experiment, irradiation of N-phthaloylglycinemethiamide <u>la</u> has produced a mixture of the expected eight- and tenmembered <u>2a</u> cyclic compounds⁴.

Irradiation of N-phthaloylglycyl-L-methionine methyl ester with a 400 W high-pressure mercury lamp in 10 mM acetone solution for 0.5 hr afforded a mixture of the cyclized products, $\underline{3}$ (46%) and $\underline{4}$ (5%)⁵. In support of the assigned structure, the cyclol $\underline{3}$ was readily converted into the dehydrated product (mp 255-6) on treatment with acid. Homologous N-substituted phthalimides $\underline{1}\underline{b}\underline{-c}$ containing amide bonds together with a ω -methylthic group were prepared and the photolysis was performed as above. The expected heteromeric cyclic peptide analogs, $\underline{2}\underline{b}$ (mp 270°, 16%)⁵ and $\underline{2}\underline{c}$ (mp 115-7°, 64%)⁵, respectively, were readily obtained as a result of the C-C bond formation between the imide carbonyl and the sulfide methyl through this extensive Norrish type II process ^{4,6}. For a typical example, the nmr spectrum of $\underline{2}\underline{c}$ had a methylene peak (3.57 δ , s, pyridine) instead of methyl in $\underline{1}\underline{c}$, and the molecular weight values determined by a vapor-pressure method and mass spectometry were 478 and 461, respectively, in agreement with the monomeric value (461), thus excluding the alternative possibility of the intermolecular

reaction. On treatment with p-toluenesulfonic acid, $\frac{2c}{c}$ similarly afforded the dehydrated compound 5 (mp 199-201; nmr, vinyl H, 6.54 δ , s, DMSO).

Number of amino acid residues in linear peptide substrates is important in determining the efficiency to give cyclic peptides^{3a}. The phthalimides with long side chains containing at least two amide bonds \(\frac{1}{2} \) are now cyclized in moderate yields to demonstrate versatility of this synthetic photoreaction affording macrocycles up to a twenty-one membered ring \(\frac{2}{2} \), which is nearly equivalent in size to that of cyclic heptapeptide. Further possible utilities of this approach for cyclic peptide studies including cylindrical models \(\frac{7}{2} \) and for more general synthet—ic application \(\frac{8}{2} \) are currently under investigation.

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References and Notes

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